

CLASS: XIIth

DATE:

SOLUTIONS

SUBJECT: CHEMISTRY

DPP NO.: 3

Topic:-HYDROCARBONS

1 (d)

Compound has $8 + 2 = 10\pi$ electrons hence is aromatic. has $4\pi e^-$, has $8\pi e^-$, while

has
$$8 + 1 = 9\pi e^-$$
, hence all these species are not aromatic

It is a mixture of solid hydrocarbons.

3 (c)

$$C_6H_5$$
 $C-CN$
 C_6H_5
 $C-CN$
 C_6H_5
 CN
 C_6H_5
 CN
 CN
 CON
 CON

2,3-diphenyl-

1,3-butadiene

This reaction is an example of Diel's Alder reaction

4 (d)

All of these can be used in cracking.

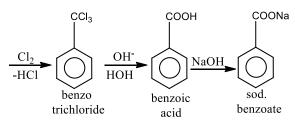
5 **(b**)

General formula of a cycloalkane is C_nH_{2n} .

6 (h)

Toluene reacts with excess of Cl_2 in presence of sunlight, the last product of this reaction is benzotrichloride which on hydrolysis gives benzoic acid, and it gives sodium benzoate on reaction with NaOH.

$$\begin{array}{c|c} \mathsf{CH}_3 & \mathsf{CH}_2\mathsf{Cl} & \mathsf{CHCl}_2 \\ \hline \\ \mathsf{toluene} & \mathsf{Cl}_2 & \mathsf{Denzal} \\ \hline \\ \mathsf{toluene} & \mathsf{chloride} \\ \end{array}$$



8 **(d**

 $\mathsf{CH_3CH_2COOH} + \mathsf{NaHCO_3} \longrightarrow \mathsf{CH_3CH_2COONa} + \mathsf{H_2O} + \mathsf{CO_2}$



Smart DPPs

9 **(a)**

 $C_2H_2 + \frac{5}{2}O_2 \longrightarrow 2CO_2 + H_2O$; 1 mole or 1 vol. of C_2H_2 requires 2.5 mole or 2.5 vol. of O_2 .

10 (a

Conjugate dienesare mores stable than the other dienes.

11 (c)

Branched chain alkanes give rise to increase on octane no.

12 (d)

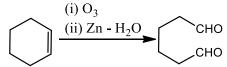
Follow Markownikoff's rule.

13 **(b)**

HOCl has Cl⁺ and OH⁻ ions

$$CH_{3}CH = CH_{2} + CI^{+} \xrightarrow{\text{Electrophilic} \\ \text{addition}} CH_{3} - CH_{-}CH_{2} - CI \\ \downarrow OH^{-} \\ CH_{3} - CH_{-}CH_{2} - CI \\ CH_{3} - CH_{2} - CH_{2} - CI \\ CH_{3} - CH_{2} - CH_{2} - CI \\ CH_{3} - CH_{3}$$

14 **(b)**



 $Zn - H_2O$ is the reagent for reductive work up of ozonide. $H_2O_2 - CH_3COOH$ would give $HOOC - (CH_2)_4 - COOH$.

15 (d

The π -bond is unshared in electromeric effect to give +ve and -ve centres on molecule.

$$CH_2 \xrightarrow{C} CH_2 \xrightarrow{A.R.} CH_4 \xrightarrow{C} CH_4$$

16 (d)

Tropylium cation is planar and have 6π -electron according to Huckel rule, hence it is

Tropylium cation is planar and have 6π -electron according to Huckel rule, hence it is aromatic.

Cyclopentadienyl anion is planar and have 6π -electron, hence it is also aromatic compound.

17 (a)

Follow peroxide effect.

18 **(a)**

In the laboratory, nitrobenzene is prepared by nitration of benzene with the mixture of nitric acid and sulphuric acid at temperature below 60° C. In which HNO₃ acts as a base

19 **(d)**

The reaction is ozonolysis. During the reaction C=C breaks to give carbonyl compounds.



$$CH_3CH = CH_2 \xrightarrow[H_2O_2]{O_3} CH_3CHO + HCHO$$

Acetaldehyde formaldehyde

20 (a)

Petrol or gasoline contains mainly C_6 to C_{11} atoms liquid alkanes.

ANSWER-KEY										
Q.	1	2	3	4	5	6	7	8	9	10
A.	D	В	C	D	В	В	D	D	A	A
						1		1		
Q.	11	12	13	14	15	16	17	18	19	20
A.	С	D	В	В	D	D	A	A	D	A
			A	17			2			

SMARTLEARN COACHING