





Solutio

Class : XIIth Date : Subject : CHEMISTRY DPP No. : 3

Topic :- Biomolecules

3	(c)
	Oils are unsaturated esters or glycerides olein is ester of unsaturated acid.
4	(b)
	Aldehydes and α -hydroxyl ketones give positive Tollen's test. Glucose is a polyhydroxy aldehyde
	and fructose is an α -hydroxyl ketone
5	(b)
J	
	Lysine contains two basic groups.
c	e.g., NH ₂
6	
-	Glucose is $CHO(CHOH)_4CH_2OH$.
7	(b)
	C_1 carbon of monosaccharides is called anomeric carbon. When the – OH group attached with C_1
	carbon is towards right, it is called α – from and when the – OH group is towards left, it is called
	β –from. Such pair of optical isomers which differ in the configuration only around anomertic
	carbon are called anomers.
	H OH HO C H aomeric carbon
	C (asymmetric)
	aomeric
	carbon (asymmetric) a-form B -form
8	(c)
0	
	Glucose $\xrightarrow{\text{Conc.H}_2\text{SO}_4}$ 6C + 6H ₂ O; this is dehydration.
9	
	Reserved fat ac <mark>t as the</mark> rmoinsulator.
12	
	Both surfactants and detergents possess the surface activity, i.e., the tendency lower surface
	tension of water. A surfactant also having cleansing action, i. e., detergency in addition to surface
	activity is called detergent.
14	(c)
	Sucrose gives glucose and fructose on hydrolysis with invertase enzyme.
	$C_{12}H_{22}O_{11} + H_2O \xrightarrow{\text{Invertase}} C_6H_{12}O_6 + C_6H_{12}O_6$
	Sucrose glucose fructose
	0
15	(c)
	A characteristic of detergent.
16	(a)
	The general formula of saturated acids is $C_n H_{2n} O_2$ or $C_n H_{2n+1} COOH$.
17	(c)
	The two polynucleotide chains or strands of DNA are joined by hydrogen bonding between the
	nitrogenous base molecules of their nucleotide monomers
19	(d)
10	



(d)

A nucleotide contains a pentose sugar [deoxyribose (in DNA) or ribose (in RNA)], nitrogenous base [such as adenine or guanine or thymine (in DNA) or cytosine or uracil (in RNA)] and a phosphate molecule.

Ising

20

At pH = 4, an amphoteric Zwitter ion structure changes into cation when an acid is added to it.

 $R \xrightarrow{\text{CH}} CH \xrightarrow{\text{COO}} \xrightarrow{\text{HH}^+} R \xrightarrow{\text{CH}} CH \xrightarrow{\text{COOH}} COOH$ $\circledast |_{\text{NH}_3} \xrightarrow{\text{HH}^+} R \xrightarrow{\text{CH}} H_3$

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Smart DPPs

ANSWER-KEY												
Q.	1	2	3	4	5	6	7	8	9	10		
A.	D	С	С	B	B	D	B	С	C	С		
Q .	11	12	13	14	15	16	17	18	19	20		
А.	С	В	Α	C	C	Α	С	В	D	D		

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