









It suggests polymerization on the lost of vinylic hydrogen atom, which is not possible.

8 (d)

Electron withdrawing groups make the monomer more reactive towards anionic polymerization

Thiokol is polymer of CH₂ClCH₂Cl and sodium polysulphide Na—S—S—Na and thus, not 9 (c) polydiene rubber.

11 (c)

The process of condensation polymerization takes place in the following manner.

Condense A - BA + BMonomers dimer

 $A - B + A \xrightarrow{\text{Condense}} A - B - A$ $A \longrightarrow B \longrightarrow A + B \longrightarrow A \longrightarrow B \longrightarrow A \longrightarrow B$

In this process no initiator is required and it is also called step growth polymerization.

Glycerol trinitrate is explosive. 13 (c)

14 (c)

Polyhydroxy butyrate — CO— β —hydroxyl valerate(PHBV) is a biodegradable polymer.

16 (b)

Saran is a copolymer of vinyl chloride and vinyledene chloride.

17 (b)

Nylon-6, 6, is polyamide having —CONH gp. 0

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 $(N - (CH_2)_6 - N - C - (CH_2)_4 - C_{n} is nylon-6,6.$

18 (a)

Melamine plastic crockery is a copolymer of HCHO and



(a) 20

Teflon is used for this purpose.

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ANSWER-KEY										
Q.	1	2	3	4	5	6	7	8	9	10
А.	A	A	D	A	А	С	D	D	С	В
)
Q.	11	12	13	14	15	16	17	18	19	20
А.	С	А	С	С	А	В	В	А	В	А